

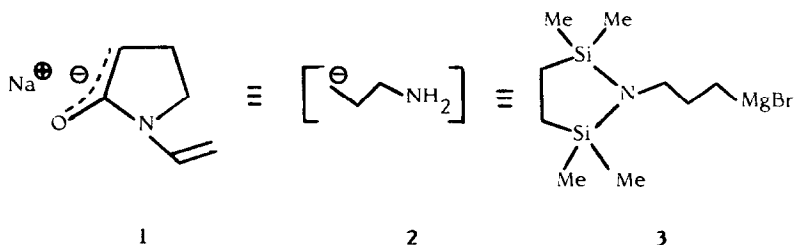
**CHELATION ASSISTED TRANSMETALLATION OF TETRAALKYL TIN DERIVATIVES :  
 C-METALLATED LITHIOCARBAMATES AS  $d^3$  REAGENTS  
 FOR THE SYNTHESIS OF  $\delta$ -HYDROXYCARBAMATES.**

Ahmed ZIDANI and Michel VAULTIER \*

Groupe de Recherches de Physicochimie Structurale, U.A. n° 704, Campus de Beaulieu,  
 Avenue du Général Leclerc, 35042 RENNES CEDEX (FRANCE).

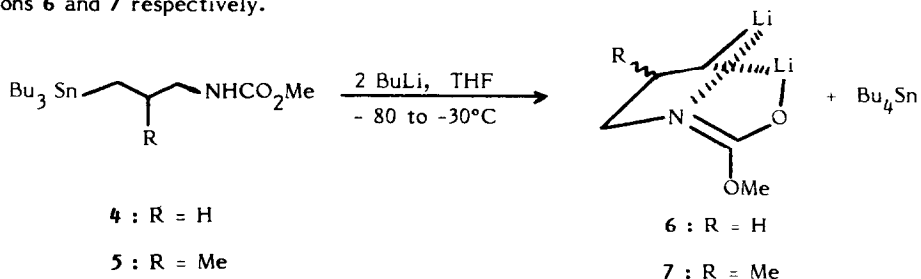
**Abstract** - The  $\gamma$ -stannylcarbamates **4** and **5** were subjected to metallation with two equivalents of BuLi to give the dianions **6** and **7** acting as  $d^3$  reagents. The C-metallation involves a chelation assisted transmetallation.

Primary amino protected Grignard or lithio reagents that can be easily formed are scarce<sup>(1)</sup>. Recently, the sodium enolate of N-vinylpyrrolidone<sup>(2)</sup> **1** and the Grignard reagent<sup>(3)</sup> **3** were used as synthetic equivalents of the synthon **2**.



We wish to report herein on the formation of a new synthetic equivalent of the synthon **2** and its reactions with some electrophiles.

The  $\gamma$ -stannylcarbamates **4** and **5**<sup>(4)</sup> were subjected to metallation with two equivalents of BuLi (2,0 M in hexane) in THF at low temperature to give rapidly (1 hour) and quantitatively the dianions **6** and **7** respectively.



The hydrolysis of solutions of **6** and **7** with saturated  $\text{NH}_4\text{Cl}$  gave quantitatively (G.C. Yields  $\cong 99\%$ ) the N-methoxycarbonyl propylamine and N-methoxycarbonylisobutylamine respectively and the tetrabutyltin. The dianion **6** was C-methylated with dimethylsulfate leading to the N-methoxycarbonylbutylamine in an 71% isolated yield. No alkylation could be achieved with alkylhalides ( $\text{ICH}_3$ ,  $\text{Br}-\text{CH}_2\text{CH}_2\text{CH}_3$ ). The dianions **6** and **7** react with carbonyl derivatives to give the protected amino-alcools **8** with good yields.

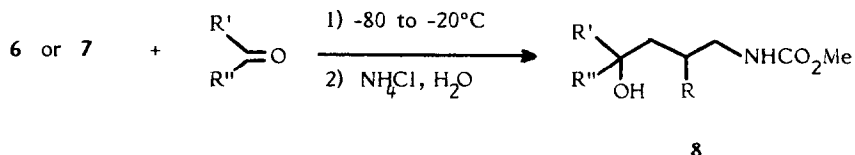




Table - Synthesis of the  $\delta$ -hydroxycarbamates **8** <sup>(5)</sup>

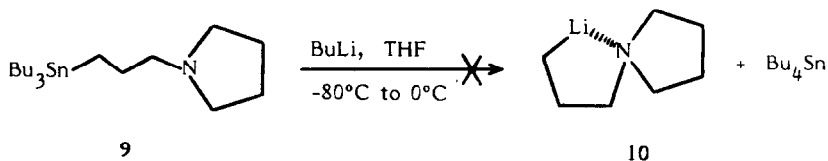
N°	R	R'	R''	Eb <sub>mm/Hg</sub> °C or m.p. °C	Yield % (a)
<b>8 a</b>	H	Ph	H	80-81	72
<b>8 b</b>	H	Et	H	100-105/0,01	72
<b>8 c</b>	H		H	58-59	61
<b>8 d</b>	H	Ph	Ph	105-106	80
<b>8 e</b>	H	Me	Me	100-103/0,01	68
<b>8 f</b>	H	-(CH <sub>2</sub> ) <sub>5</sub> -		74-75	71
<b>8 g</b>	H	-(CH <sub>2</sub> ) <sub>4</sub> -CH-   CH <sub>3</sub>		67-68	74 (b)
<b>8 h</b>	H		H	(c)	70
<b>8 i</b>	Me	Ph	H	120-122/0,01	80 (d)
<b>8 j</b>	Me	Ph	Ph	108-109	80

(a) Yields are of isolated pure products, after column chromatography and distillation or recrystallization. (b) A single diastereoisomer (<sup>13</sup>C NMR) of unknown stereochemistry was obtained. (c) oil, TLC : R<sub>f</sub> = 0,45 (ether/methanol 8/2). (d) 1/1 mixture of two diastereoisomers.

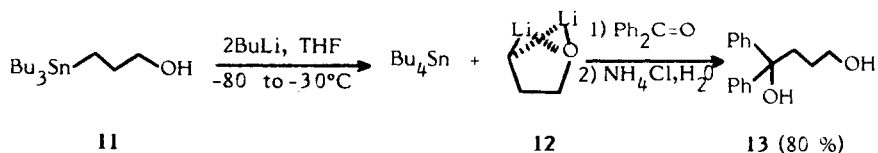
The very easy transmetalation of the tetraalkyltin derivatives **4** and **5** is intriguing. It has been shown that transmetalation of tetraalkyltin compounds with alkylolithiums would give complex equilibrium mixtures <sup>(6)</sup> except in special cases <sup>(7)</sup>.

Recently, B.K. Carpenter and R.H. Newman-Evens<sup>(8)</sup> have published a stereoselective synthesis of bicyclo [2-1-1] hexenes-5-d using a transmetallation of a tetraalkyltin assisted by chelation with an OLi group. These authors suggest the possibility of having a tin "ate" complex instead of a dianion. In order to get some insight into this problem, we have recorded the  $^{119}\text{Sn}$  NMR spectrum<sup>(9)</sup> of the solution obtained by treatment of **4** with 2 BuLi at  $-20^\circ\text{C}$ . A single sharp singlet was observed at  $-11.3$  ppm. This spectrum was not modified by the addition of one equivalent of pure  $\text{Bu}_4\text{Sn}$ . Therefore, we believe that, in our case, there is only the formation of the dianion **6** and  $\text{Bu}_4\text{Sn}$ . The presence of an anion in  $\delta$  position to the tin atom seems to be necessary to shift the transmetallation equilibrium towards the formation of the dianions **6** and **7**<sup>(10)</sup>. This stems from the following experiments.

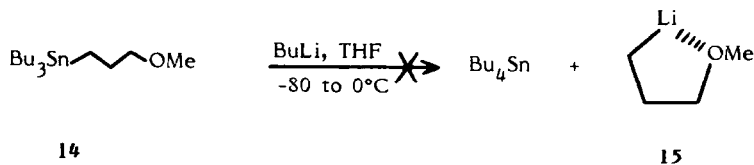
The addition of one equivalent of BuLi to the tetraalkyltin **9** did not produce any anion **10** nor  $\text{Bu}_4\text{Sn}$  as it could be seen from the GC analysis of the hydrolyzed crude reaction mixture. **9** was quantitatively recovered.



Furthermore, the treatment of **11** with two equivalents of BuLi (2,0 M in hexane) gave the dianion **12** which was added to benzophenone leading to the diol **13**.



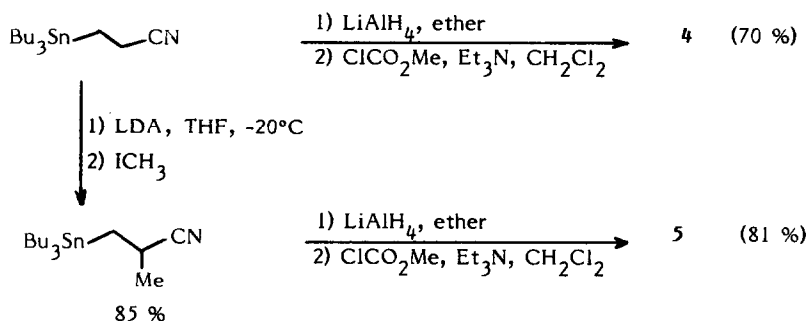
The anion **15** could not be generated from **14** in the indicated conditions.



Finally, we believe in agreement with B.K. Carpenter and R.H. Newman-Evens<sup>(8)</sup> and in a more general way that it could be possible to use a suitably disposed negatively charged atom to direct a tetraalkyltin transmetalation and thus to obtain a variety of new useful nucleophilic reagents.

## References and Notes

- 1 - J.C. STOWELL, *Chem. Rev.*, 1984, 84, 409.
- 2 - P. JACOB III, *J. Org. Chem.*, 1982, 47, 4165.
- 3 - F.Z. BASHA and J.F. DEBERNARDIS, *Tetrahedron Letters*, 1984, 25, 5271.
- 4 - The tetraalkyltin reagents **4** and **5** were prepared from  $\beta$ -tri-n-butylpropionitrile (A.J. Leusing and J.G. Noltes, *Tetrahedron Letters*, 1966, p. 335) according to the following scheme :



- 5 - All new compounds gave satisfactory analytical data for C and H and spectra ( $^1\text{H}$ ,  $^{13}\text{C}$ , IR and mass) in accord with the proposed structures.
- 6 - D. SEYFERTH and M.A. WEINER, *J. Am. Chem. Soc.*, 1962, 84, 361.
- 7 - D.J. PETERSON, *J. Am. Chem. Soc.*, 1971, 93, 4027. D.J. PETERSON and J.F. WARD, *J. Organomet. Chem.*, 1974, 66, 209.  
W.C. STILL, *J. Am. Chem. Soc.*, 1978, 100, 1481.  
E.J. COREY and T.M. ECKRICH, *Tetrahedron Letters*, 1984, 25, 2415.  
R. GOSWAMI and D.E. CORCORAN, *Tetrahedron Letters*, 1982, 23, 1463.
- 8 - R.H. NEWMAN-EVANS and B.K. CARPENTER, *Tetrahedron Letters*, 1985, 26, 1141.
- 9 -  $^{119}\text{Sn}$  NMR spectra (proton decoupled) were recorded at 111,9 MHz.  $\text{Me}_4\text{Sn}$  was used as the reference (capillary tube).
- 10 - The treatment of the aminostannanes  $\text{Bu}_3\text{Sn}-(\text{CH}_2)_3-\text{NH}_2$  and  $\text{Bu}_3\text{Sn}-(\text{CH}_2)_3-\text{NHSi}(\text{Me})_3$  respectively with three and two equivalents of BuLi in the aforementioned conditions leads to the quantitative formation of  $\text{Bu}_4\text{Sn}$  and therefore indicating that the transmetallation is also possible for these substrates. The chemistry of the corresponding anions has not yet been studied.

(Received in France 11 December 1985)